

5. Quantum Mechanics (Spring 2005)

An electron moves in a hydrogen atom potential - ignoring spin and relativity - in a state  $|\psi\rangle$  that has the wave function

$$\psi(r, \theta, \phi) = NR_{21}(r) [2iY_1^{-1}(\theta, \phi) + (2+i)Y_1^0(\theta, \phi) + 3iY_1^1(\theta, \phi)]$$

where the  $Y_l^m(\theta, \phi)$  are the spherical harmonics,  $R_{nl}(r)$  are the normalized hydrogen atom wave functions, and  $N$  is a positive real number.

- Calculate  $N$ .
- What is the expectation value of  $L_z$ ? ( $\hbar\mathbf{L} = \mathbf{r} \times \mathbf{p}$ )
- What is the expectation value of  $L^2$ ?
- What is the expectation value of the kinetic energy in terms of  $\hbar, c$ , the electron charge  $e$  or the fine-structure constant  $\alpha$ , and the electron mass  $m$ ?

Note: The explicit forms of the functions that appear in  $\psi(r, \theta, \phi)$  above are

$$R_{21}(r) = \frac{1}{2\sqrt{6}} \frac{r}{a^{5/2}} e^{-r/2a} \quad Y_1^{\pm 1}(\theta, \phi) = \mp \sqrt{\frac{3}{8\pi}} \sin(\theta) e^{\pm i\phi} \quad Y_1^0(\theta, \phi) = \sqrt{\frac{3}{4\pi}} \cos(\theta)$$

a. By normalization,  $1 = \langle \Psi | \Psi \rangle = \int_{-\infty}^{\infty} \int \Psi^*(r, \theta, \phi) \Psi(r, \theta, \phi) r^2 dr d\Omega$

$$= |N|^2 \int_0^{\infty} R_{21}^*(r) R_{21}(r) r^2 dr (4 + 5 + 9) \quad \text{since } \int (Y_l^m)^* Y_l^m d\Omega = \delta_{ll'} \delta_{mm'}$$

$$= |N|^2 \int_0^{\infty} \frac{1}{24} \frac{r^4}{a^5} e^{-r/a} dr$$

$$= \frac{3}{4} |N|^2 a^{-5} (a^5 4!) = 18 |N|^2 \Rightarrow N = \frac{1}{\sqrt{18}} \quad \text{up to a phase}$$

b.  $|\Psi\rangle = \frac{2i}{\sqrt{18}} |21-1\rangle + \frac{2+i}{\sqrt{18}} |210\rangle + \frac{3i}{\sqrt{18}} |211\rangle$   
 and  $L_z |nlm\rangle = \hbar m |nlm\rangle$

$$\Rightarrow \langle \Psi | L_z | \Psi \rangle = \frac{4}{18} \langle 21-1 | L_z | 21-1 \rangle + \frac{5}{18} \langle 210 | L_z | 210 \rangle + \frac{9}{18} \langle 211 | L_z | 211 \rangle = -\frac{4}{18} \hbar + \frac{9}{18} \hbar = \frac{5}{18} \hbar$$

c.  $L^2 |nlm\rangle = \hbar^2 l(l+1) |nlm\rangle$

$$\Rightarrow \langle \Psi | L^2 | \Psi \rangle = \frac{4}{18} \langle 21-1 | L^2 | 21-1 \rangle + \frac{5}{18} \langle 210 | L^2 | 210 \rangle + \frac{9}{18} \langle 211 | L^2 | 211 \rangle = \frac{8}{18} \hbar^2 + \frac{10}{18} \hbar^2 + \frac{18}{18} \hbar^2 = 2\hbar^2$$

d. We know the total energy because this is a Hydrogen atom in an  $n=2$  energy eigenstate, so  $E = -\frac{me^4}{2\hbar^2(2)^2} = -\frac{me^4}{8\hbar^2}$   
 So we can use the virial theorem to get the kinetic energy

$$\langle T \rangle = -\langle E \rangle = \frac{me^4}{8\hbar^2}$$