

14. *Statistical Mechanics and Thermodynamics* (Spring 2005)

A photon gas in thermal equilibrium is contained within a box of volume V at temperature T .

- Use the partition function to find the average number of photons \bar{n}_r in the state having energy E_r .
- Find a relationship between the radiation pressure P and the energy density u (i.e. the average energy per unit volume).
- If the volume containing the photon gas is decreased adiabatically by a factor of 8, what is the final pressure if the initial pressure is P_0 ?

$$\begin{aligned} \text{a. } Z &= \sum_r e^{-\beta E_r} = \sum_r e^{-\beta (n_1 \epsilon_1 + n_2 \epsilon_2 + \dots)} \\ &= \sum_{n_1=0}^{\infty} \sum_{n_2=0}^{\infty} \dots e^{-\beta (n_1 \epsilon_1 + n_2 \epsilon_2 + \dots)} \\ &= \left(\sum_{n_1=0}^{\infty} e^{-\beta n_1 \epsilon_1} \right) \left(\sum_{n_2=0}^{\infty} e^{-\beta n_2 \epsilon_2} \right) \dots \\ &= \left(\frac{1}{1 - e^{-\beta \epsilon_1}} \right) \left(\frac{1}{1 - e^{-\beta \epsilon_2}} \right) \dots \end{aligned}$$

$$\Rightarrow \ln(Z) = \sum_{r=1}^{\infty} \ln \left(\frac{1}{1 - e^{-\beta E_r}} \right) = - \sum_{r=1}^{\infty} \ln(1 - e^{-\beta E_r})$$

$$\text{Therefore } \bar{n}_r = -\frac{1}{\beta} \frac{\partial \ln(Z)}{\partial E_r} = +\frac{1}{\beta} \frac{1}{1 - e^{-\beta E_r}} \beta e^{-\beta E_r} = \frac{1}{e^{\beta E_r} - 1}$$

$$\begin{aligned} \text{b. } P &= \frac{1}{\beta} \frac{\partial \ln(Z)}{\partial V} = -\frac{1}{\beta} \sum_{r=1}^{\infty} \frac{1}{1 - e^{-\beta E_r}} \beta \frac{\partial E_r}{\partial V} e^{-\beta E_r} \\ &= - \sum_{r=1}^{\infty} \frac{1}{e^{\beta E_r} - 1} \frac{\partial E_r}{\partial V} = - \sum_{r=1}^{\infty} \bar{n}_r \frac{\partial E_r}{\partial V} \end{aligned}$$

$$\text{Now } E_r = \hbar K C = \hbar C \sqrt{\left(\frac{n_x \pi}{L}\right)^2 + \left(\frac{n_y \pi}{L}\right)^2 + \left(\frac{n_z \pi}{L}\right)^2} = \frac{\hbar C \pi}{V^{1/3}} \sqrt{n_x^2 + n_y^2 + n_z^2}$$

$$\Rightarrow \frac{\partial E_r}{\partial V} = -\frac{1}{3} \frac{\hbar C \pi}{V^{4/3}} \sqrt{n_x^2 + n_y^2 + n_z^2} = -\frac{E_r}{3V}$$

$$\Rightarrow P = \sum_{r=1}^{\infty} \bar{n}_r \frac{E_r}{3V} = \frac{1}{3V} \sum_{r=1}^{\infty} \bar{n}_r E_r = \frac{E}{3V} = \frac{1}{3} u$$

c. $dE = -pdV$ since $dQ = 0$ for an adiabatic process

$$\Rightarrow d(uV) = -\frac{4}{3} dV$$

$$\Rightarrow u dV + V du = -\frac{4}{3} dV$$

$$\Rightarrow \frac{dV}{V} + \frac{du}{u} = -\frac{4}{3} \frac{dV}{V}$$

$$\Rightarrow \frac{du}{u} = -\frac{4}{3} \frac{dV}{V}$$

$$\Rightarrow \ln(u) = -\frac{4}{3} \ln(V) + C$$

$$\Rightarrow u = A V^{-4/3}$$

$$\Rightarrow P = A' V^{-4/3} \text{ so this is a polytropic process}$$

$$\Rightarrow P_0 V_0^{4/3} = P_f V_f^{4/3} = A' \Rightarrow P_f = P_0 \left(\frac{V_0}{V_f}\right)^{4/3} = P_0 (8)^{4/3} = 16 P_0$$